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equation for our first
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equation.***

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begin to solve
problems using the
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equation that we just
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video we introduce
the concept of a
stoichiometric table,
which is an essential
tool for solving
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of Rate of Reaction

is very important to

understand. $-r_A$ is

essentially: how

many moles are

being reacted ...

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efficiency in problem
5 b to $.42 \times .7 =$
.294. My apologies
on that silly mistake!***

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video, we have

discussed the

Enzymatic , Kinetics

, using the Michaelis-

Menten model and

this will help the

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